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5-Arylazoimidazo[2,1-b]-1,3,4-Thiadiazoles—New Dichroic T-Dyes For LCD

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INTRODUCTION

Continuing research of dichroic *T*-dyes^{1–4} we have obtained new 5-arylazo-2,6-diarylimidazo[2,1-b]-1,3,4-thiadiazoles (5,9–12). The compounds (2–5) have been synthesized by known methods.^{5,6}

The distinctive feature of these *T*-dyes is their sufficiently high geometric anisotropy and more symmetrical arrangement of the arylazogroup in comparison to the substituted 4-aryl-5-arylazo-1,2-diaminoimidazoles earlier described by us.⁴ Quantum-chemical calculations of the latter⁷ and dichroism data of their long-wave absorption band (LAB) in liquid crystals⁴ showed that oscillator of electron transition (which is responsible for LAB) is nearly perpendicular to their long geometric axis. It was supposed for this reason that oscillator of long-wave electron transition in new *T*-dyes (5,9–12) would be also nearly perpendicular to their long geometric axis because 4-aryl-5-arylazo-1,2-diaminoimidazoles and their derivatives as well as *T*-dyes (5,9–12) contain fragments (13,14 respectively) which are similar by electron structure and are responsible for LAB.

Consequently it was possible to expect that *T*-dyes (5,9–12) in liquid crystals exhibit a high value of negative dichroism.



EXPERIMENTAL

The compounds (2-12) were synthesized by known methods. Data of elemental analysis were satisfactory. Individuality of synthesized compounds was checked by TLC. Polarized spectra of *T*-dye solutions (5,9-12) in liquid crystalline mixture of 4-alkyl- and 4-alkoxy-4'-cyanobiphenyls with clearing point equal to 60°C were measured at 20°C by an earlier described method.⁸ Dichroism values were calculated on the base of D_{\parallel} and D_{\perp} values by formula $S = (D_{\parallel} - D_{\perp}) / (D_{\parallel} + 2D_{\perp})$

DISCUSSION

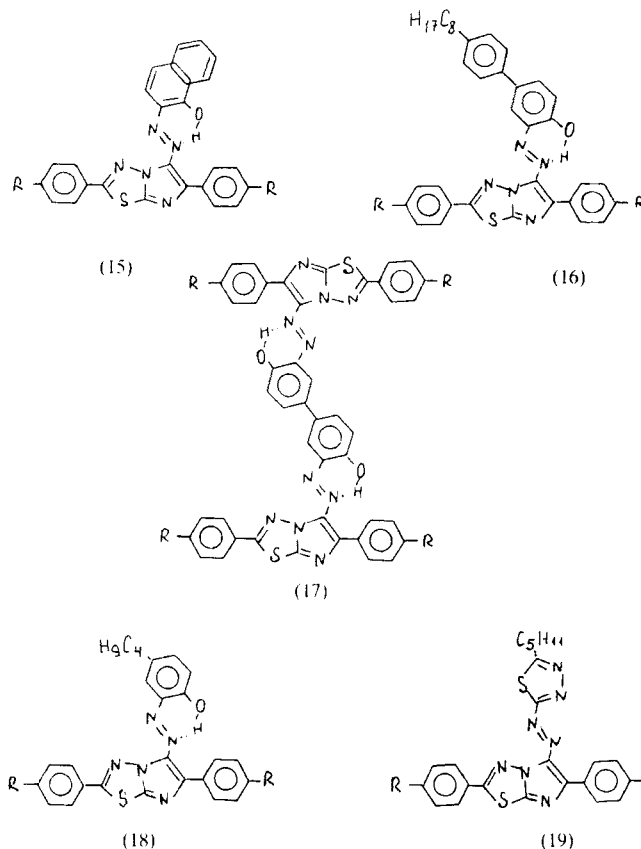
T-dyes (5,9-12) have LAB at 460–530 nm and $S = -0.23 + -0.31$ (Table).

As follows from the Table, the *T*-dye (10) with 2-hydroxynaphthyl-azogroup and the bis-azodye (12) have the longest absorption band that correlates with well-known data. It is necessary to note that the azodye (12) is the first representative of a bis-azodye of the *T*-type. Sufficiently high value of negative dichroism testifies that obtained *T*-dyes (5,9-12) have the structure proposed on the scheme but not alternative structures (15-19). The structures (15-19) have a lower geometric anisotropy (ratio of the length of *T*-dye molecule to its width) in comparison with the structures (5,9-12); in the case of realization of structures (15-19) lower values of negative dichroism would be expected.

It is necessary to note that the *T*-dye (11) having two rod-like fragments similar to LC molecules binded with a “cross” azogroup

TABLE I
Position of long-wave absorption band and dichroism of *T*-dyes (5,9-12)

Comp. N	M.p., °C	λ_{\max} , nm	S
5	133–135	460	–0.23
9	131–133	465, 486	–0.29, –0.29
10	175–177	488, 510	–0.27, –0.27
11	162–164	476	–0.31
12	291–293	499, 530	–0.27, –0.28



has a higher S value (-0.31) than the T -dye (9) that has only one rod-like fragment for which $S = -0.29$. The T -dye (12) has a geometrical anisotropy about twice less than the T -dye (9). Nevertheless values of their negative dichroic ratios are approximately equal. We explain this by the fact that the bis-azodye (12) has three rod-like fragments bound with azogroups. Analogous dependence of a negative dichroic ratio value on the structure of T -dyes was observed by us earlier for anthraquinone¹⁻³ and aryazoazole⁴ T -dyes. In our opinion the presence of two or more rod-like fragments in the T -dye molecule which are bound with azogroup or other fragment permitting an electron cross transition of $\pi-\pi^*$ type stimulates a dye molecular packing in the liquid crystal (Figure).

This fact as well as our earlier idea of dichroic dye design is confirmed by experimental data of this and previous works.^{1,4,10,11}

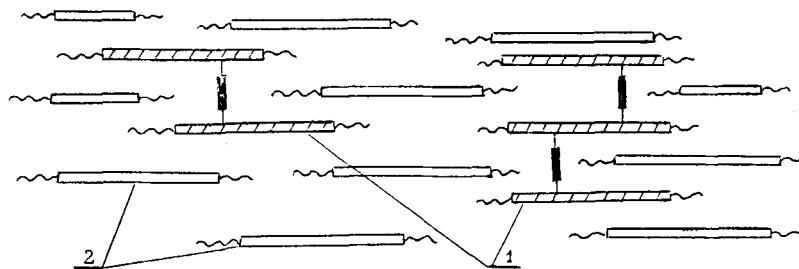


FIGURE Packing of T-dye molecules (1) including several rod-like fragments bound with transversal chromophore systems in a liquid crystal (2).

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